

ES 202

Fluid and Thermal Systems

Lecture 21:
Isentropic Efficiencies
(1/30/2003)

Assignments

- Homework:
 - 7-87, 7-90 in Cengel & Turner
 - Find rate of S_{gen} in 7-63 (Monday)
 - Find ΔS for 7-35 (no modification on Tuesday)
- Reading assignment
 - 8-5 to 8-7, 8-10, 8-11 in Cengel & Turner
 - ES 201 notes

Announcements

- Problem session this evening at 7 pm
- Short quiz next Monday on Week 7 materials

Road Map of Lecture 21

- Overview of what you have already learned
 - property changes
 - general substances versus models
 - constant versus variable specific heats (solution methods)
- Finish up example on adiabatic versus non-adiabatic turbine
- More examples
 - reinforce what was taught
 - introduce the limit of best performance
 - notion of isentropic efficiency
- Representation of isentropic and non-isentropic processes on T - s diagram

A Check List for You

- Property changes:
 - internal energy, enthalpy, entropy
 - general substances (State Principle) versus models
 - general substance: State Principle (exact differential)
 - definition of specific heats c_p , c_v
 - Gibbs equation (two different forms)
 - models: ideal gas, incompressible substance (resulting simplifications)
 - constant versus variable specific heats
 - different solution methods:
 - constant specific heats (analytical solutions)
 - variable specific heats:
 - direct integration (the hard way)
 - use property table (u , h , s^o , P_r , v_r) – the easier way
 - “average” specific heats (may require iterations if temperatures are unknown)
- Special case of isentropic process

In-Class Exercises

- Continue example on adiabatic versus non-adiabatic turbine
- Example on entropy generation

Isentropic Efficiency

- Central theme: the reversible, adiabatic (*i.e.* isentropic) process sets the **limit of best performance** of any system
- It also sets the **reference** for the definition of efficiency of compressors and turbines
- For compressors and pumps, work is done **on the system**:

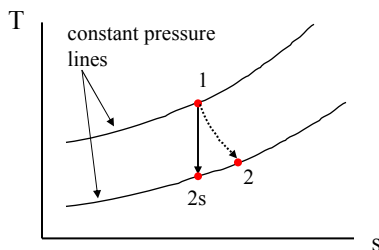
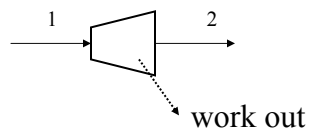
$$\eta_{comp} = \frac{\dot{W}_{isentropic}}{\dot{W}_{actual}}$$

- For turbines, work is done **by the system**:

$$\eta_{turb} = \frac{\dot{W}_{actual}}{\dot{W}_{isentropic}}$$

Representation on T - s Diagram

- Turbine



- Compressor

